

## Electrochemical Data on Compartmentation into Cell Wall, Cytoplasm, and Vacuole of Leaf Cells in the CAM Genus *Kalanchoë*

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**Summary.** Transcellular electrical profiles of *Kalanchoë* leaf cells were obtained by pushing a glass micro-saltbridge through cells with the tip consecutively in the cell wall, cytoplasm, and vacuole. The electrical resistance of the cell wall was too small to be detectable, that of the plasmalemma and tonoplast was about 0.18–0.21 and 0.16–0.18  $\Omega\text{m}^2$ , respectively. The electrical potential difference between the cytoplasm and the external medium,  $\psi_{co}$ , was  $\sim -180\text{ mV}$ , the potential difference between the vacuole and the medium,  $\psi_{vo}$ , was  $\sim -155\text{ mV}$ , and thus the mean potential difference at the tonoplast,  $\psi_{vc}$ , was about  $+25\text{ mV}$ . Potential difference,  $\psi_{vo}$ , was independent of proton concentration in the external medium between pH 9 and 5.5, and behaved like an  $\text{H}^+$ -electrode between pH 5 and 3. Depolarizations and hyperpolarizations of  $\psi_{vo}$  obtained by increasing and decreasing, respectively, the  $\text{Na}^+$ -concentrations in the medium were smaller than with changing  $\text{K}^+$ -concentrations, suggesting that permeabilities are  $P_{\text{Na}^+}/P_{\text{K}^+} \simeq 0.23$ . Assessment of  $\text{K}^+$ -compartmentation by flux analysis gave  $\text{K}^+$ -concentrations in the cytoplasm including chloroplasts ( $c_c$ ) and vacuole ( $c_v$ ) as  $c_c$  between 200 and 400  $\text{mmol kg}^{-1}\text{ FrWt}$  and  $c_v \simeq 15\text{ mmol kg}^{-1}\text{ FrWt}$ . The Nernst criterion suggests that metabolically regulated  $\text{K}^+$  transport out of the vacuoles concentrates  $\text{K}^+$  in the cytoplasm. Fusicocin ( $10^{-5}\text{ M}$ ) hyperpolarized  $\psi_{co}$  by about 100 mV and depolarized the positive  $\psi_{vc}$  by about 10 mV, the latter presumably being an insignificant effect. The evidence for the existence of proton pumps exchanging  $\text{H}^+$  and  $\text{K}^+$  at the plasmalemma and at the tonoplast is discussed.

The most outstanding feature of crassulacean acid metabolism is the occurrence of large diurnal oscillations of malic acid levels in the leaf cells. Under experimental conditions the amplitude of these oscillations in *Kalanchoë* plants can amount to up to 180  $\text{mmol malic acid per kg FrWt}$  for the 12-hr period between the end of the light phase (malic acid level: 20–40  $\text{mmol kg}^{-1}\text{ FrWt}$ ) and the end of the dark phase (200–220  $\text{mmol kg}^{-1}$ ). Feedback inhibition of the malate producing system, whose major component is phosphoenol pyruvate carboxylase, by the product malate (*cf.* Kluge & Ting, 1978) and cytoplasmic pH-control (Lüttge, 1979) are important factors in regulation of these oscillations. Energy-dependent transport of malic acid from the cytoplasm into the vacuole clearly is involved. It has been suggested that a proton pump at the tonoplast directed towards the vacuole is driving this process (Lüttge & Ball, 1979).

Levels of inorganic ions, e.g.,  $\text{K}^+$ ,  $\text{Na}^+$ ,  $\text{Cl}^-$ , in extracts of CAM leaves do not oscillate during the diurnal CAM rhythm (Lüttge & Ball, 1974). It is possible, however, that compartmentation of inorganic ions in CAM leaf cells participates in regulation of malic acid oscillations. In particular  $\text{K}^+$  and  $\text{Mg}^{2+}$  are cations whose concentration in the cytoplasm and near the tonoplast could considerably influence enzymic reactions synthesizing and breaking down malate and a putative  $\text{H}^+$ -pumping ATPase at the tonoplast, respectively. The surprisingly low  $\text{K}^+$  content of *Kalanchoë* leaves (below 20  $\text{mmol kg}^{-1}\text{ FrWt}$ ) has long been a puzzling observation (Lüttge & Ball, 1974, 1979; Lüttge, Ball & Tromballa, 1975), and it has been suggested that a mechanism must be operative concentrating  $\text{K}^+$  in the cytoplasm (Lüttge, 1979). Otherwise cytoplasmic reaction systems would be hampered by  $\text{K}^+$  deficiency (WynJones, Brady & Spears, 1979).

Among other parameters, electrical characteristics

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of the tonoplast and plasmalemma (i.e., potential differences and resistances, responses to ion gradients such as those of  $H^+$ ,  $K^+$ ,  $Na^+$ ), intracellular compartmentation of  $K^+$ , and  $K^+$  fluxes at the tonoplast and plasmalemma need to be known to evaluate these possibilities. Such data are provided in the present report.

## Materials and Methods

### Plants

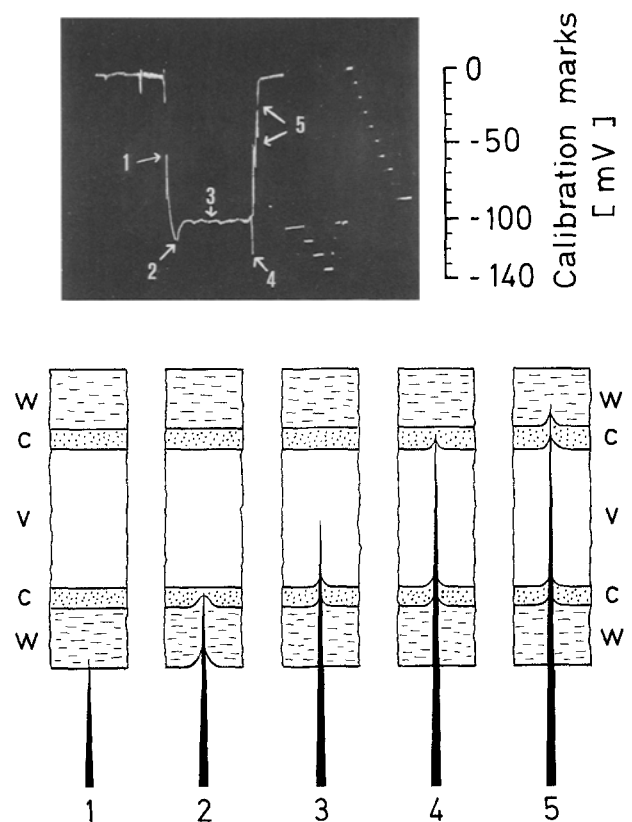
Plants of *Kalanchoë daigremontiana* Hamet et Perrier and *Kalanchoë tubiflora* (Harv.) Hamet were grown in soil culture in a glass house. Ion contents in the leaves were 10–20 mM  $K^+$ , 1–15 mM  $Na^+$ , and 10–20 mM  $Cl^-$  (Lüttge & Ball, 1979, and new analyses in the present study).

### Electrical Measurements

For measurements of electrical membrane parameters several kinds of preparations were used. In experiments with material designated "intact leaves" in Table 1, whole leaves were cut from the plants and glass micro-electrodes were inserted into cells without further slicing. In the experiments of Tables 1 and 4 and Fig. 1, leaf slices were  $2 \times 5$  mm large pieces of tissue with the cuticles removed on both surfaces. These small tissue blocks were used either shortly (i.e., 5–20 min) after preparation or after a washing period of 60–90 min at 25°C in the light. Leaf slices of the experiments shown in Figs. 3–5 were  $5 \times 10$  mm large with the cuticles left in place. Young plantlets in Table 1 were adventitious plantlets that had emerged from mature leaves. The second about 5 mm long leaves of 4 to 6 months old plantlets were used without slicing.

All preparations were kept in experimental solutions whose composition is indicated in the experiments of Figs. 3–5 in the legends. In the experiments of Fig. 1 and Table 1 the external medium contained 1 mM  $K^+$ , 1 mM  $Na^+$ , 0.25 mM  $Ca^{2+}$ , 0.25 mM  $Mg^{2+}$ , all as  $Cl^-$  salts. In Table 4 the solution was distilled water because fusicoccin was dissolved in distilled water, but similar results could be obtained with a salt solution as indicated above. Replacing this salt solution by distilled water caused a hyperpolarization of only  $-15$  to  $-20$  mV as compared to a hyperpolarization of  $-100$  mV due to fusicoccin.

For measurements of electrical membrane potentials glass micro-saltbridges (micro-electrodes) filled with 3 M KCl and connected via Ag/AgCl contacts to an electrometer and recording devices (oscilloscope, line recorder) were applied in the usual way (Lüttge & Ball, 1979; Rona, 1973; Rona et al., 1980). The reference saltbridge was in the external solution excepting experiments with intact leaves. In the latter case the reference electrode was in contact with an agar bridge (8% agar-agar, plus 1 mM KCl, 1 mM NaCl, 0.25 mM  $CaCl_2$  and 0.25 mM  $MgCl_2$ ) touching the leaf surface. For the recordings of transcellular electrical profiles the micro-electrode was pushed slowly through the cell wall, cytoplasm, and vacuole and out of the cell through the cytoplasm and cell wall at the site opposite to electrode entry (Fig. 1) with the aid of an automatic micromanipulator (model MM6 of Oskar Uhl, Asslar-Wetzlar, W-Germany; Rona, Cornel & Heller, 1977). The electrode also could be kept for up to 5 min in the cytoplasm. In other cases the electrode was inserted directly into the vacuole. In this way potentials were obtained which often were stable for many hours and the effects of various parameters could be in-



**Fig. 1.** Example of a transcellular electrical profile. This profile was obtained by the electrode penetrating a cell in a freshly cut leaf slice (i.e., 5 min after preparation) of *K. daigremontiana*, where absolute values of electrical potentials are comparatively low (see Table 1):  $\psi_{wo}$  (1,5) is  $-40$  to  $-48$  mV;  $\psi_{co}$  (2,4) is  $-110$  to  $-122$  mV;  $\psi_{vo}$  (3) is  $-90$  to  $-100$  mV. The first  $\psi_{co}$ -value (2) is smaller than the second one (4) and is an underestimation because the electrode entering the cell has to be pushed against the mechanic resistance of the cell wall and thus passes too rapidly through the cytoplasm into the vacuole. This problem does not occur upon exit of the electrode from the vacuole into the cytoplasm so that the second value (4) is more reliable (Rona et al., 1977)

vestigated (i.e., Figs. 3–5). Measurements of electrical resistance were made with the "one-electrode" technique (Brenneke & Lindemann, 1971; Anderson, Henderson & Higinbotham, 1974; Rona & Cornel, 1979; Rona, 1979) since it was extremely difficult to insert more than one electrode into the *Kalanchoë* leaf cells. A small constant current (4 nA, square signal of a frequency of 10 kHz) was injected into the cells via the same glass micro-electrode serving for measurement of membrane potentials. The difficulties of this method have been discussed by Etherton, Keifer and Spanswick (1977). Results obtained here are similar to those found in the literature for other plant materials (Walker, 1960; Greenham, 1966; Spanswick, 1970, 1972; Anderson et al., 1974; Rona & Cornel, 1979). Furthermore, for purposes of the present investigation, the relative resistances were more important than the correct absolute values. Resistance measurements were made under the same conditions as measurements of membrane potentials; often they were made simultaneously. The increase in resistance as the electrode passed from the cell wall into the cytoplasm and then into the vacuole and the decrease as the electrode was pushed out of the cell was an important control for identification of compart-

ments during measurements of transcellular electrical profiles in addition to microscopic observation of the localization of the tip of the electrode (Rona et al., 1980).

### Potassium Flux Analysis

Analysis of samples of 2-mm wide leaf slices suspended in 5 mM KCl plus 0.1 mM CaSO<sub>4</sub> showed that there was very little change in K<sup>+</sup> content of the tissue over a period of about 36 hr. The rate of net K<sup>+</sup> uptake was small (see Fig. 2) and for the purpose of calculation of fluxes was taken to be zero. In this case the fluxes at the plasmalemma ( $\phi_p$ ) and tonoplast ( $\phi_t$ ) can be determined as follows (Walker & Pitman, 1976):

a) Tissue was put in labeled solution for 16 hr and then transferred to unlabeled solution of otherwise identical composition to measure the rate of tracer loss from the tissue. The labeling solution was 5 mM KCl + 0.1 mM CaSO<sub>4</sub>, and <sup>86</sup>Rb was used as a tracer for K<sup>+</sup> at a specific activity of approximately 0.5 mCi/mmol K<sup>+</sup>. The substitution of <sup>86</sup>Rb for K was checked by comparison of <sup>42</sup>K and <sup>86</sup>Rb uptakes from the same solution (Fig. 8).

b) When uptake and elution are at the same temperature, fluxes can be calculated from the analysis of the efflux curve (as in Fig. 6) into exponential components. The basis for these calculations is given in Walker and Pitman (1976). During uptake over a period  $t'$  the amount of tracer in the tissue will rise to

$$Q_T = A(1 - e^{-k_1 t'}) + B(1 - e^{-k_2 t'}).$$

During elution to unlabeled solution  $Q_T'$  will fall as

$$Q_T' = A(1 - e^{-k_1 t'})e^{-k_1 t} + B(1 - e^{-k_2 t'})e^{-k_2 t}.$$

Analysis of the rate of loss of tracer from the tissue and the decrease in  $Q_T'$  can be used to estimate the rate constants  $k_1$ ,  $k_2$ ; the amounts  $Q_c' = A(1 - e^{-k_1 t'})$  and  $Q_v' = B(1 - e^{-k_2 t'})$ ; the rate of tracer loss at  $t=0$  ( $\Phi_{co}$ ) and the rate of tracer loss in the slow component extrapolated to  $t=0$  ( $\Phi_{vo}$ ). In this case,  $A$  and  $B$  can be calculated from  $Q_c'$  and  $Q_v'$  as  $t'$  is known, and  $\phi_p$  and  $Q_T$  calculated since

$$A = (\phi_p - k_2 Q_T) / (k_1 - k_2)$$

$$B = (k_1 Q_T - \phi_p) / (k_1 - k_2).$$

This calculated value of  $Q_T$  can be compared with  $Q_T$  from chemical analysis to give a check on the method. The cytoplasmic content  $Q_c$  can be estimated from the relationship that  $(k_1 + k_2) = \phi_p / Q_c + Q_T k_1 k_2 / \phi_p$ . The flux  $\phi_t$  can be calculated since

$$\phi_p / \phi_t = \frac{\Phi_{co} - \Phi_{vo}}{\Phi_{vo}} \cdot \bar{S}_v \text{ (see below)}$$

and

$$\bar{S}_v \text{ is } Q_v' / Q_v = Q_v' / (Q_T - Q_c).$$

c) When uptake is at a different temperature from efflux this simple (and exact) calculation is not valid. However, during the efflux we can write

$$\Phi_{co} = \phi_p \cdot \bar{S}_c = \frac{\phi_p (\phi_p + \phi_t \cdot \bar{S}_v)}{\phi_p + \phi_t}$$

where  $\bar{S}_c$  is the specific activity in the cytoplasm at the end of the uptake and

$$\Phi_{vo} = \phi_p \cdot \bar{S}_{ce} = \phi_p \frac{(\phi_t \cdot \bar{S}_v)}{\phi_p + \phi_t}$$

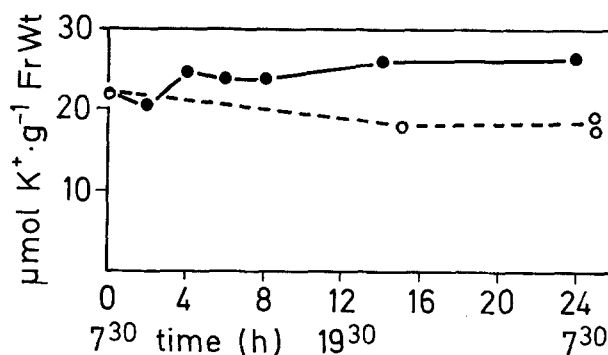


Fig. 2. K<sup>+</sup> levels in *K. daigremontiana* leaf slices which were pre-treated in 5 mM KCl plus 0.1 mM CaSO<sub>4</sub> for 15 hr then transferred to fresh solution at time zero (7<sup>30</sup>) at 21 °C (closed circles) and 6 °C (open circles). Samples were taken, blotted dry, and analyzed for K<sup>+</sup> as indicated

where  $\bar{S}_v$  is the vacuolar specific activity at the end of the uptake (start of the elution) and  $\bar{S}_{ce}$  is the specific activity during elution extrapolated to  $t=0$ .

Hence

$$\phi_p / \phi_t = \frac{\Phi_{co} - \Phi_{vo}}{\Phi_{vo}} \cdot \bar{S}_v, \text{ and } \phi_p / (\phi_p + \phi_t)$$

can be determined and substituted with  $\bar{S}_v$  in  $\Phi_{vo}$  to give  $\phi_t$ . The content of the cytoplasm,  $Q_c$ , can be calculated from  $k_1 = (\phi_p + \phi_t) / Q_c$  (which is true when  $Q_c \ll Q_v$  and  $\phi_t \ll Q_v$ ). The value of  $\bar{S}_v$  needed in these calculations can be that calculated for the end of the uptake period at 25° using the method b) above. The calculated value of  $Q_T$  at 6° then uses the estimate of  $Q_c$  at 6° plus  $Q_v$  at the end of the uptake period.

Elutions were made at 6 and at 25 °C. Separate measurements of tracer uptake were performed by adding tissue to labeled solution and extracting samples at intervals. Each sample was rinsed for 2 min in ice cold, unlabeled solution, blotted and weighed, and then counted.

Stock solutions of <sup>86</sup>Rb (RbCl in aqueous solution, 1–12 mCi/mg Rb) and of <sup>42</sup>K (target material KCl) were obtained from Amersham Buchler, Braunschweig. Counting was performed with a methane gas flow counter on planchets.

Potassium levels in tissue samples were determined by flame photometry of aqueous extracts.

### Sampling of Leaf Material for Experiments in Relation to the Diurnal Rhythm of CAM.

It was observed earlier that the electrical potential difference between the vacuole of cells in *K. daigremontiana* leaf slices and the medium,  $\psi_{vo}$ , is independent of the phase of the diurnal rhythm of CAM at which the leaves were harvested (Lüttge & Ball, 1979). Total K<sup>+</sup> levels in the tissue do not change during the rhythm (Lüttge et al., 1975). K<sup>+</sup> and Cl<sup>-</sup> uptake by *K. daigremontiana* leaf slices also is independent of the phase of the rhythm (Lüttge & Ball, 1974). Thus in the present attempt to investigate compartmentation of K<sup>+</sup> and electrical properties of plasmalemma and tonoplast, it seemed justified to pay no attention to the phase of the rhythm at which the plant material was sampled. In the experiments of flux analysis requiring a loading period of 15 hr plus a washout period of 24 hr or more this would have been useless anyhow. Sampling usually was during the light phase; in the experiments of Fig. 1 and Tables 1 and 4 it was between 2 and

6 p.m. This is a first approach only and more rapid methods are needed to establish ion levels in the various compartments in relation to the CAM rhythm, i.e., to test the possibility of diurnal oscillations in the ion compartmentation.

#### Symbols and Abbreviations:

$c$  = concentration  
 CAM = crassulacean acid metabolism  
 FC = fusicoccin  
 FrWt = fresh weight  
 FS = free space  
 HEPES = *n*-hydroxy-ethyl-piperazine-N'-2-ethanesulfonic acid  
 $k$  = rate constant  
 MES = 2-(*n*-morpholino)-ethanesulfonic acid  
 $N$  = Nernst  
 $P$  = permeability coefficient  
 $Q$  = quantity, content  
 $Q'$  = amount of tracer  
 $R$  = electrical resistance  
 $r$  = specific electrical membrane resistance  
 $S$  = specific activity  
 $t$  = time

Used as subscripts only (two symbols in series indicate a direction):

$c$  = cytoplasm  
 $i$  = inside the cells  
 $o$  = external medium  
 $p$  = plasmalemma  
 $t$  = tonoplast

$T$  = total of  $c$  and  $v$

$v$  = vacuole

$w$  = cell wall

Greek symbols:

$\psi$  = electrical potential difference at a membrane

$\phi$  = flux

$\Phi$  = experimental measure of tracer exchange

## Results and Discussion

### Transcellular Electrical Profiles

Figure 1 depicts the gradual penetration of a *Kalanchoë* cell with a glass micro-electrode and gives an example for a corresponding transcellular electrical profile. Table 1 summarizes data obtained from numerous such experiments.

The potential obtained by penetrating the cell wall is rather independent of the material and conditions (Table 1). Similar results were obtained with *Lupinus luteus* roots (*unpublished*) and isolated cells of *Acer pseudoplatanus* L. (Rona et al., 1980). The electrical resistance of the cell wall is too small to be detected with the methods used; this is expected in view of the high permeability of the cell wall which is

**Table 1.** Electrical potential differences and resistances obtained from transcellular electrical profiles of *K. daigremontiana* and *K. tubiflora* leaf cells under various conditions<sup>a</sup>

Materials and conditions	$w \rightarrow o$	$c \rightarrow o$				$v \rightarrow o$			$v \rightarrow c$		
	$\psi$ [mV]	$\psi$ [mV]	$R$ [M $\Omega$ ]	$r$ [ $\Omega\text{m}^2$ ]		$\psi$ [mV]	$R$ [M $\Omega$ ]	$r$ [ $\Omega\text{m}^2$ ]	$\psi$ [mV]	$R$ [M $\Omega$ ]	$r$ [ $\Omega\text{m}^2$ ]
<i>Kalanchoë daigremontiana</i> :											
Intact leaves	$-45 \pm 6$ (9) -30 to -60 (80)	$-183 \pm 11$ (13) -170 to -221 (80)	$7 \pm 2$ (13) 3.5 to 10 (80)	0.18		$-159 \pm 7$ (15) -130 to -198 (80)	$13 \pm 2$ (12) 11 to 19 (80)	0.34	+24	6	0.16
Leaf slices 5 to 15 min after preparation	$-36 \pm 11$ (8)	$-90 \pm 8$ (13)				$-64 \pm 7$ (11)			+26		
Leaf slices 60 to 90 min after preparation	$-38 \pm 10$ (9)	$-174 \pm 7$ (12)				$-153 \pm 9$ (17)			+21		
Young plantlets	$-31 \pm 12$ (8)	$-120 \pm 12$ (13)				$-95 \pm 14$ (10)			+25		
<i>Kalanchoë tubiflora</i> :											
Intact leaves	$-33 \pm 7$ (6) -20 to -40 (6)	$-167 \pm 9$ (15) 4 to 12 (12)	$8 \pm 2$ (12) 4 to 12 (12)	0.21		$-141 \pm 5$ (12)	$15 \pm 2$ (19) 13 to 20 (19)	0.39	+26	7	0.18

<sup>a</sup> Averages with SD and numbers of replicates in brackets and in some cases ranges of individual measurements are given. Cell wall resistances ( $R_{wo}$ ) were below detectable values. Specific resistances,  $r$ , were calculated on the basis of an average diameter of 0.091 mm ( $\pm 0.014$  SD) of the approximately spherical cells of *K. daigremontiana* leaves measured on sections of 266 cells (by courtesy of D. Kramer and J.A.C. Smith). This may underestimate the specific resistance in cells of a complex leaf tissue, because due to the electrical coupling between individual cells (Spanswick, 1972; Drake, 1979) a large surface may be in effect.

readily accessible to ions and thus is usually considered as free space (Grignon & Lamant, 1973; Rona, 1979).

Resistance increases markedly upon penetration into the cytoplasm. Although the role of the cell wall should not be generally neglected (Heller, Grignon & Rona, 1974; Demarty, Ripoll & Thellier, 1980), the electrical potential difference between the cytoplasm and the medium ( $\psi_{co}$ ) can be considered to be largely a plasmalemma potential.  $\psi_{co}$  is slightly but not significantly larger in cells of intact leaves of *K. daigremontiana* than of *K. tubiflora*. Cells of young 4–6 months old plantlets obtained from vegetative propagation on leaves of older plants have lower  $\psi_{co}$  than those of adult leaves. This agrees with earlier observations where  $\psi_{co}$  of cells in adult leaves was about  $-200$  mV on average but cells in the leaflets of the small adventitious plantlets growing between the dents of mature leaves only had  $\psi_{co}$  of  $-126$  mV (Lüttge & Ball, 1979). Slicing much reduces  $\psi_{co}$  but recovery is already observed after about 1 hr of washing. Membrane potential changes due to slicing and recovery due to “adaptive ageing” (Van Steveninck, 1976) are well known from a variety of plant materials including leaves (see Lüttge & Ball, 1979, and references therein).

Resistance increases further as the tip of the electrode passes into the vacuole. The electrical potential between the vacuole and the medium ( $\psi_{vo}$ ) is the sum of gradients across the tonoplast ( $\psi_t$ ) and plasmalemma plus cell wall ( $\psi_{co}$ ). The differences between the two *Kalanchoë* species and the effects of slicing, “adaptive ageing” and leaf age discussed above for  $\psi_{co}$  are also reflected in  $\psi_{vo}$ .

The electrical potential difference at the tonoplast can be estimated as

$$\psi_t = \psi_{vo} - \psi_{co}.$$

Table 1 shows that for all conditions and materials  $\psi_t$  is very similar. This is an important result, suggesting that regulation of the overall electrical potential difference,  $\psi_{vo}$ , resides at the plasmalemma, changed conditions such as age, slicing, and “adaptive ageing” affecting  $\psi_{co}$  but not  $\psi_t$ . This agrees with experiments on cells of the moss *Mnium cuspidatum*, where depolarization by cyanide was exclusively due to plasmalemma depolarization (Fischer, Lüttge & Higinbotham, 1976). Electrogenicity resides largely at the plasmalemma. This reflects itself also in the finding that potentials at the tonoplast are only very small and positive.

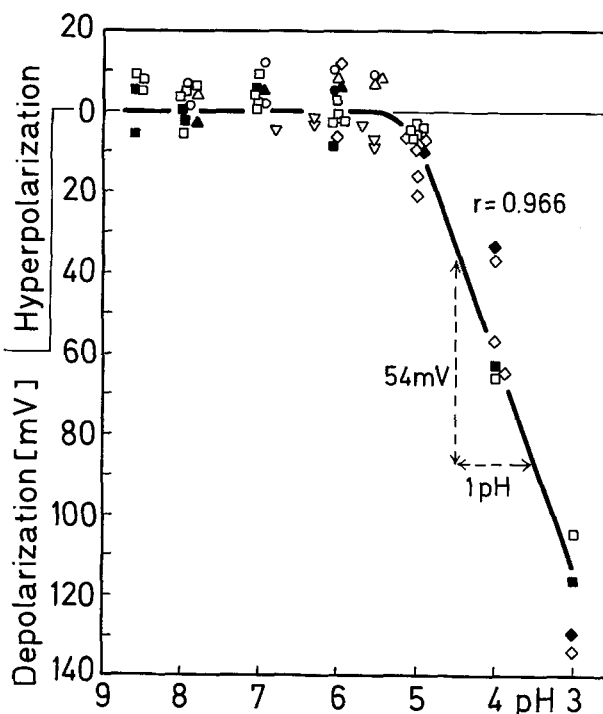
The electrical resistance of the plasmalemma  $R_p$  is equal to  $R_{co}$  because  $R_{wo}$  is not detectable, and

$$R_t = R_{vo} - R_{co}.$$

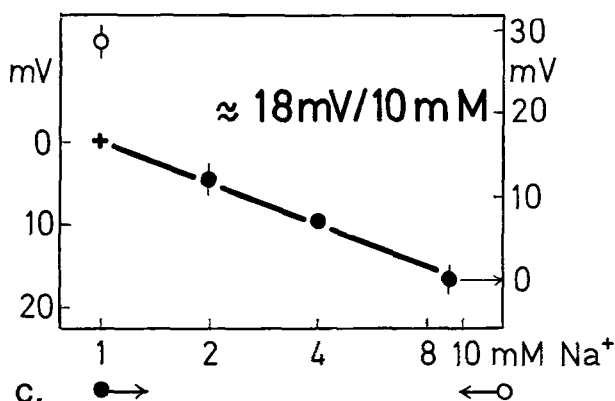
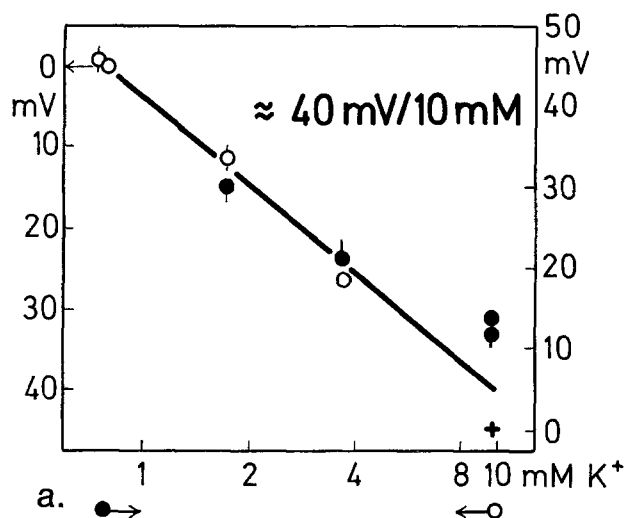
The specific resistances of the plasmalemma and tonoplast are not significantly different (Table 1).

#### Responses of $\psi_{vo}$ to Gradients of $H^+$ , $K^+$ and $Na^+$

It was demonstrated earlier that  $\psi_{vo}$  of leaf cells of *K. daigremontiana* is independent of the amount of malic acid stored in the vacuoles and hence of the pH gradients between the vacuoles and the cytoplasm or the vacuoles and the outer medium (Lüttge & Ball, 1979). Figure 3 shows that  $\psi_{vo}$  is also independent of the external pH between pH 9 and about 5.5. Below pH 5,  $\psi_{vo}$  behaves like an  $H^+$ -electrode, for which at

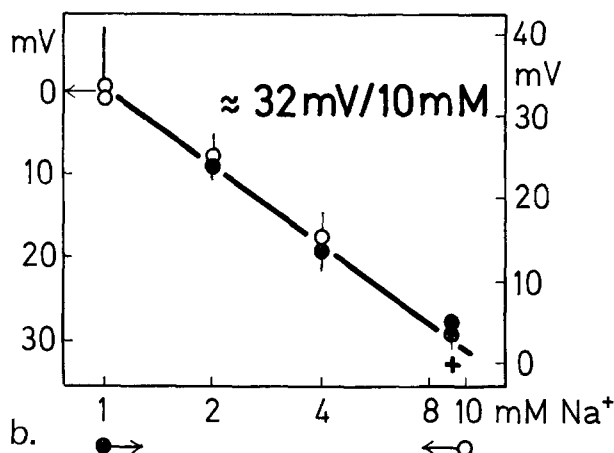


**Fig. 3.** Dependence of  $\psi_{vo}$  of cells in *K. daigremontiana* leaf slices on pH in the external medium. Leaf slices were incubated overnight at pH 6 or 7 in the same MES and HEPES buffers, respectively, which were subsequently used during the experiments. After impalement of electrodes establishment of a constant resting potential at pH 7 was attended, then the pH was changed, and the de- and hyperpolarizations obtained stepping pH up and down and back to the initial pH 7 are plotted. Different symbols denote different buffer solutions, open and closed symbols refer to experiments in the light and in the dark, respectively. All solutions contained  $0.1$  mM  $CaCl_2$ ;  $K^+$ -concentrations ( $K_2SO_4$ ) were  $0.75$  to  $0.85$  mM excepting the squares, where the medium contained no  $K^+$ .  $Na^+$  concentrations varied since the pH of the solutions was adjusted with  $NaOH$ . By additions of  $Na_2SO_4$ ,  $Na^+$  concentration was kept constant, however, within a given buffer series as follows: circles —  $10$  mM buffer,  $6.8$  mM  $Na^+$ ; upright triangles —  $1$  mM buffer,  $0.7$  mM  $Na^+$ ; inverted triangles —  $0.1$  mM buffer,  $0.13$  mM  $Na^+$ ; squares and diamonds —  $1$  mM buffer,  $1$  mM  $Na^+$ . The tissue contained  $20 \pm 3$  mmol  $K^+ \cdot kg^{-1}$  FrWt and  $2.7 \pm 0.8$  mmol  $Na^+ \cdot kg^{-1}$  FrWt (SD).



room temperature a depolarization of 58 mV per pH unit would be observed. For higher plants a similar pH dependence was obtained with *Impatiens balsamina* root cells (Jones, Novacky & Dropkin, 1975). In the ranges tested pH-response of *K. daigremontiana* was independent of the kind (MES or HEPES), strength, and  $K^+$  and  $Na^+$  content of the buffer solutions. By contrast to observations with *Nitellopsis obtusa* (Gyenes et al., 1978) the response was not affected by light.

Figures 4 and 5 show the dependence of  $\psi_{vo}$  on  $K^+$  and  $Na^+$  concentrations and ratios, respectively, in the outer medium. Unfortunately, these experiments could not be performed with cells whose electrogenic mechanisms were inhibited and which had only a diffusion potential. Due to equilibration problems of *K. daigremontiana* leaf slices  $\psi_{vo}$  responds rather sluggishly to inhibitors (Lüttge & Ball, 1979). Electrode impalements are usually lost before a stable diffusion potential is attained after addition of an inhibitor to the medium. Thus it is difficult to estimate the permeability coefficients from the data of



**Fig. 4.** Dependence of  $\psi_{vo}$  of cells in *K. daigremontiana* leaf slices on  $K^+$  and  $Na^+$  concentration in the external medium. Depolarization (●); Hyperpolarization (○). Leaf slices were preincubated overnight in solutions as indicated below. After establishment of a constant resting potential following electrode impalement in the same solution as during pretreatment (= crosses in the graphs serving as points of reference),  $K^+$  or  $Na^+$  concentrations were stepped up or down and back to the original concentration and the resulting de- and hyperpolarizations were plotted. (a): Preincubation with 0.1 mM  $CaSO_4$  and 9.6 mM  $K^+$  (as sulphate);  $K^+$  and  $Na^+$  levels in the tissue after preincubation were  $23 \pm 0.6$  and  $2.7 \pm 0.2$  mmol  $kg^{-1}$  FrWt;  $K^+$  varied in experiment. (b): Preincubation with 0.1 mM  $CaSO_4$  and 9.2 mM  $Na^+$  (as sulphate);  $K^+$  and  $Na^+$  levels in the tissue  $16 \pm 1.4$  and  $7 \pm 0.9$  mmol  $kg^{-1}$  FrWt;  $Na^+$  varied. (c): Preincubation with 0.1 mM  $CaSO_4$  and 1 mM  $Na^+$  (as sulphate);  $K^+$  and  $Na^+$  levels in the tissue  $17 \pm 3.3$  and  $3 \pm 0.1$  mmol  $kg^{-1}$  FrWt;  $Na^+$  varied. Errors are SD

Figs. 4 and 5 without further assumptions.  $R_{Na^+}/P_{K^+}$  can be calculated from the Goldman equation

$$\Delta E = -\frac{RT}{F} \ln \frac{c_i^{K^+} P_{K^+} + c_i^{Na^+} P_{Na^+}}{c_o^{K^+} P_{K^+} + c_o^{Na^+} P_{Na^+}},$$

where  $R$ ,  $T$  and  $F$  have the usual thermodynamic meanings. The assumption must be made that the condition of constant field holds for a plant tissue, for which justification may be limited. If it is further assumed that the hyperpolarization of about  $-38$  mV obtained in Fig. 5 by replacing 8 mM  $K^+$  by 8 mM  $Na^+$  is a diffusion and not an electrogenic effect and that the internal concentrations  $c_i^{Na^+}$  and  $c_i^{K^+}$  do not change during the experiment,  $R_{Na^+}/P_{K^+}$  can be calculated as about 0.23. A similar relationship, i.e.,  $R_{Na^+} < P_{K^+}$ , is obtained with algal cells (see Table 6.5 in Raven, 1976), with many plant roots (see Pitman, 1976), and with isolated cells of *Acer pseudoplatanus* (Pennarun et al., 1978), but in maize root epidermis it was found that the situation can also be inverse

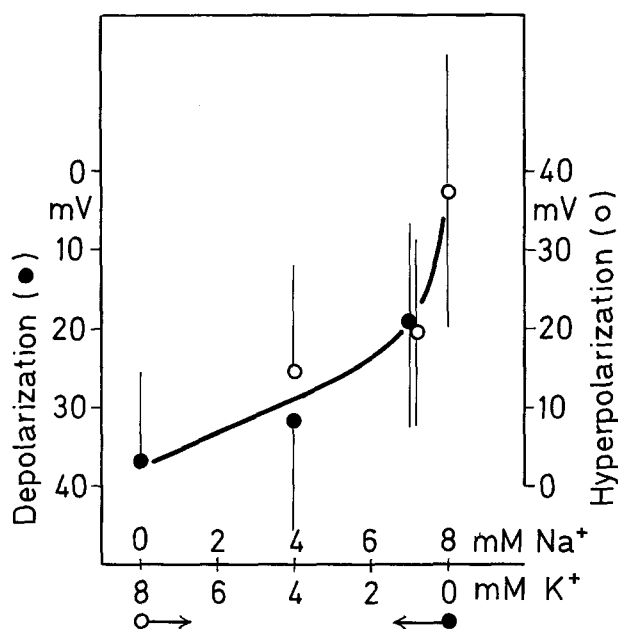


Fig. 5. Dependence of  $\psi_m$  in *K. daigremontiana* leaf slices on the  $\text{Na}^+/\text{K}^+$  ratio in the external medium. Hyperpolarizations obtained by stepwise replacing 8 mM  $\text{K}^+$  by  $\text{Na}^+$  (open symbols) and depolarizations due to stepwise replacement of  $\text{Na}^+$  by  $\text{K}^+$  (closed symbols) are plotted

depending on the  $\text{Na}^+$  and  $\text{K}^+$  concentrations in the medium (Michalov, 1977a, b).

#### Analysis of $\text{K}^+$ -Fluxes and Intracellular Compartmentation of $\text{K}^+$

Figure 6 gives an example of the tracer exchange plotted as  $\log_e$  (tracer in tissue) vs. time of elution. There was a rapidly exchanging component of about  $1.1 \mu\text{mol g}^{-1} \text{FrWt}$  with a rate constant of  $k_{FS} = 4.5 \cdot 10^{-3} \text{sec}^{-1}$  (i.e.,  $t_{1/2}$  about 150 sec) due to elution of the apoplastic free space.

Results for elution of the more slowly exchanging components at 6° and 25° for 2 experiments are given in Table 2. Data at 25° were analyzed using uptake and efflux information as outlined above (p. 27). For comparison with the calculated value of  $Q_T$ , the equivalent estimates from analysis of the tissue in the two experiments were 16 and  $17.4 \mu\text{mol g}_{\text{FrWt}}^{-1}$  (Table 2). Data at 6° were analyzed using the efflux data only together with the value of  $S_v$  from the 25° experiment. The fluxes at the plasmalemma were not altered by the temperature difference, but the fluxes into the vacuole were higher at 25° than at 6°. Estimates of the cytoplasmic content were about  $2\text{--}3 \mu\text{mol g}_{\text{FrWt}}^{-1}$  with about  $12\text{--}15 \mu\text{mol g}_{\text{FrWt}}^{-1}$  in the vacuoles.

An alternative approach to estimating  $Q_c$  and  $Q_v$

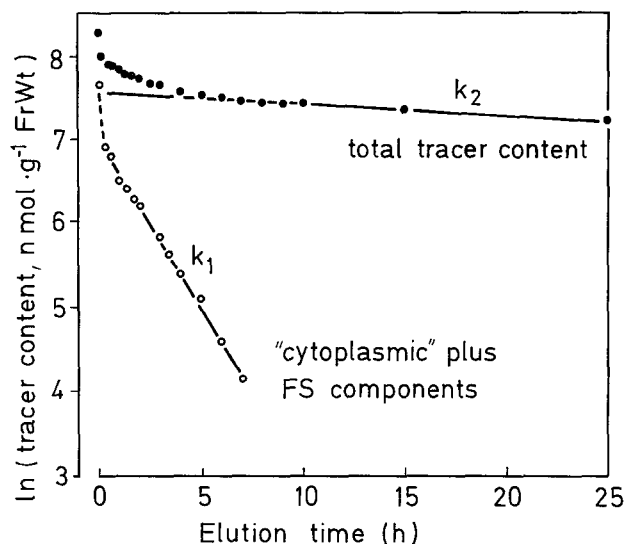


Fig. 6. Time course of tracer exchange at 6°C after loading for 16 hr in  $^{86}\text{Rb}$  labeled 5 mM KCl solution. The "cytoplasmic" plus FS components (open points) are calculated as the differences between the straight line and total tracer content (solid points). In this example  $k_1 = 1.055 \cdot 10^{-4} \text{sec}^{-1}$  and  $k_2 = 2.95 \cdot 10^{-6} \text{sec}^{-1}$

is to measure the time course of tracer uptake, which is given by the expression for  $Q_T$  given on p. 27. In Fig. 7 a comparison of uptake at 2 temperatures was fitted to uptake expressions of this type, but including a term for free space content (FS). This analysis yielded the following values for fluxes ( $\text{pmol g}^{-1} \text{sec}^{-1}$ ) and  $Q_c$  ( $\mu\text{mol g}^{-1}$ )

	$\phi_p$	$\phi_i$	$Q_c$
6°	305	40	1.0
25°	430	100	2.0

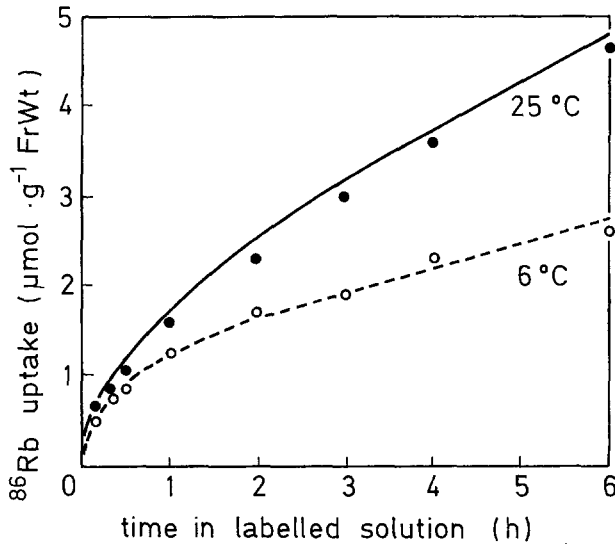
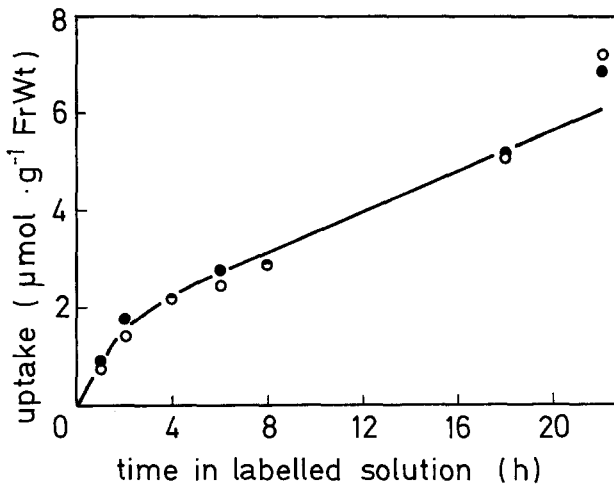
The fluxes are higher than those in Table 2, due to differences in the tissue, but the results show the same general values and effect of temperature on  $\phi_i$ . Note that there was also some effect on  $\phi_p$  and a lower value of  $Q_c$  at 6° than at 25°. Other values for fluxes are from data of Fig. 8 (also at 25°) i.e.:  $\phi_p = 235$ ,  $\phi_i = 40 \text{ pmol g}^{-1} \text{sec}^{-1}$  and  $Q_c = 1.2 \mu\text{mol g}^{-1}$ .

Figure 8 shows the validity of using  $^{86}\text{Rb}$  as a tracer for  $\text{K}^+$ , by comparing  $^{42}\text{K}$  and  $^{86}\text{Rb}$  uptakes to the same tissue.

There is thus support both from uptake and efflux measurements for a cytoplasmic  $\text{K}^+$  content of about  $2 \mu\text{mol g}^{-1} \text{FrWt}$  and vacuolar  $\text{K}^+$  of about  $15\text{--}17 \mu\text{mol g}^{-1} \text{FrWt}$ .

**Table 2.** K<sup>+</sup> fluxes and intracellular compartmentation in adult leaves of *Kalanchoë daigremontiana* based on efflux analysis

Temp. (°C)	Data						Calculated values					
	$Q'_c$	$Q'_v$	$Q_T$	$k_1$	$k_2$	$\Phi_{co}$	$\Phi_{vo}$	$\phi_p$	$\phi_t$	$Q_c$	$Q_v$	$Q_T$
	(μmol g <sup>-1</sup> FrWt)			(sec <sup>-1</sup> × 10 <sup>-6</sup> )		(pmol g <sup>-1</sup> FrWt sec <sup>-1</sup> )		(pmol g <sup>-1</sup> FrWt sec <sup>-1</sup> )		(μmol g <sup>-1</sup> FrWt)		
6	1.12	2.35	16.0	100	1.8	115	4.2	140	32	1.65	11.1	12.8
25	1.04	1.82	16.0	105	2.9	115	5.3	145	42	1.8	11.1	12.9
6	1.37	2.87	17.4	87	2.5	125	7.2	150	41	2.4	12.8	15.2
25	1.21	2.87	17.4	78	3.8	105	10.8	150	77	3.0	12.8	15.8

**Fig. 7.** Uptake of <sup>86</sup>Rb from 5 mM KCl + 0.1 mM CaSO<sub>4</sub> at 25 °C (closed circles) and 6 °C (open circles). The lines are calculated using the relation of uptake = FS + A(1 - e<sup>-k<sub>1</sub>t</sup>) + B(1 - e<sup>-k<sub>2</sub>t</sup>) (see text). Values of A and B were 1.0 × 10<sup>-6</sup> mol g<sup>-1</sup> FrWt and 17.0 × 10<sup>-6</sup> mol g<sup>-1</sup> FrWt at 25 °C and 0.7 × 10<sup>-6</sup> mol g<sup>-1</sup> FrWt and 17.3 × 10<sup>-6</sup> mol g<sup>-1</sup> FrWt at 6 °C, and k<sub>1</sub>, k<sub>2</sub> were 263 × 10<sup>-6</sup> sec<sup>-1</sup> and 9.8 × 10<sup>-6</sup> sec<sup>-1</sup> at 25 °C and 333 × 10<sup>-6</sup> sec<sup>-1</sup> and 4.2 × 10<sup>-6</sup> sec<sup>-1</sup> at 6 °C**Fig. 8.** Uptake of <sup>86</sup>Rb (open circles) and <sup>42</sup>K (closed circles) at 25 °C from 5 mM KCl and 0.1 mM CaSO<sub>4</sub>. The line is calculated (as in Fig. 7) using A = 0.7 × 10<sup>-6</sup> mol g<sup>-1</sup>; B = 17.3 × 10<sup>-6</sup> mol g<sup>-1</sup>; k<sub>1</sub> = 235 × 10<sup>-6</sup> sec<sup>-1</sup> and k<sub>2</sub> = 3.97 × 10<sup>-6</sup> sec<sup>-1</sup>

### K<sup>+</sup> Electrochemical Gradients at the Plasmalemma and Tonoplast

Results obtained by electrical measurements (Table 1) and compartmental analysis (Table 2) can be used to evaluate K<sup>+</sup>-electrochemical gradients at the plasmalemma and tonoplast, of adult *K. daigremontiana* leaves, e.g., by the Nernst criterion

$$E_{N,p}^{K^+} = -\frac{RT}{F} \ln \frac{c_c^{K^+}}{c_o^{K^+}}; E_{N,t}^{K^+} = -\frac{RT}{F} \ln \frac{c_v^{K^+}}{c_c^{K^+}}.$$

The relative volume of the cytoplasm is needed for calculation of  $c_c$  and  $c_v$  from  $Q_c$  and  $Q_v$ , respectively. (Note that  $Q_c$  and  $Q_v$  are *contents* in the compartments  $c$  and  $v$  on a total tissue FrWt basis, whereas  $c_c$  and  $c_v$  are *concentrations*.) On electronmicrographs of *K. daigremontiana* leaves D. Kramer (*unpublished*) has estimated the volume of the cytoplasm including chloroplasts as 0.5–1 % of the total cell volume. Table 3 compares K<sup>+</sup> Nernst potentials obtained on this basis with the measured electrical potential differences. Unfortunately the electrical and the flux measurements were made in two different laboratories at external K<sup>+</sup>-concentrations of 1 and 5 mM, respectively. Calculations for both alternatives are given in Table 3. The calculations suggest that metabolic energy is needed to transport K<sup>+</sup> out of the vacuole into the cytoplasm and out of the cytoplasm to the medium. It should be noted that for the intact (aerial) leaves the plasmalemma fluxes are not so relevant, a large volume of external solution being absent. Experiments with *K. daigremontiana* leaf slices incubated for 15 hr in various KCl concentrations suggest that the tissue may be in equilibrium with a 10 mM K<sup>+</sup> solution because the net change of K<sup>+</sup> in the leaf slices was zero in this solution (Fig. 9). (This was also the reason for choosing for the flux analyses an external K<sup>+</sup> concentration of 5 mM, which is close enough to the equilibrium concentration – see Fig. 9 – and still low enough to allow sufficient specific radioactivity of the labeling solutions.)

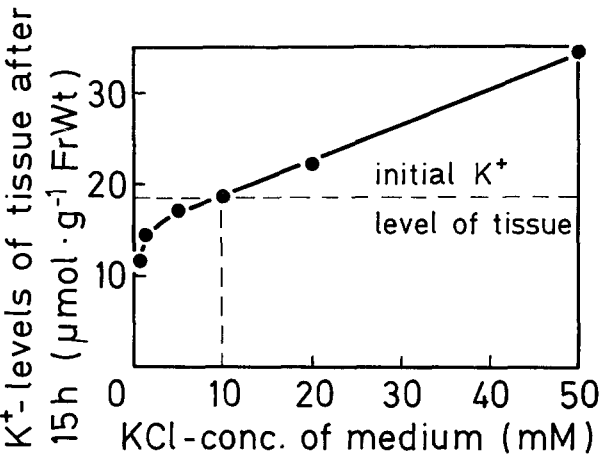
Thus it appears that metabolically regulated K<sup>+</sup> transport out of the vacuoles concentrates K<sup>+</sup> in the



**Table 3.** K<sup>+</sup>-electrochemical gradients at the plasmalemma and tonoplast of adult leaves of *Kalanchoë daigremontiana*

$Q_c$	$Q_v$	$c_o$	$c_c$	$c_v$	Plasmalemma		Tonoplast	
					$E_{N,p}^{K+}$	$\psi_p$	$E_{N,t}^{K+}$	$\psi_t$
[mmol kg <sup>-1</sup> FrWt]		[mM]			[mV]		[mV]	
2	15	5	400 <sup>a</sup>	15	-112	~-180	+84	+25
2	15	1	400 <sup>a</sup>	15	-154	~-180	+84	+25
2	15	5	200 <sup>b</sup>	15	-95	~-180	+66	+25
2	15	1	200 <sup>b</sup>	15	-136	~-180	+66	+25

<sup>a</sup> Cytoplasm=0.5% of cell volume      <sup>b</sup> 1% cytoplasm.



**Fig. 9.** Changes of K<sup>+</sup> levels of *K. daigremontiana* leaf slices after 15 hr incubation in KCl solutions of varied concentration (+0.1 mM CaSO<sub>4</sub>) at 21 °C in the dark

cytoplasm, so that sufficiently large levels are attained in spite of the very low overall K<sup>+</sup> content of *Kalanchoë* leaf cells.

#### The Effects of Fusicoccin (FC) on the Plasmalemma and the Tonoplast

Table 4 shows that FC hyperpolarizes  $\psi_{co}$  and  $\psi_{vo}$ . This confirms earlier results showing an FC-elicited hyperpolarization of  $\psi_{vo}$  of *K. daigremontiana* leaf cells (Lüttge & Ball, 1979). The hyperpolarization of  $\psi_{co}$  of about 100 mV is consistent with the literature, suggesting that all higher plant cells have an electrogenic proton extrusion pump at the plasmalemma which is specifically stimulated by FC (Marré, 1977, 1979). In the transcellular electrical profiles obtained with FC it was very difficult to distinguish between cytoplasm and vacuole, and thus it may be preferable to consider the FC effect somewhat less specifically as an “internal” hyperpolarization. Nevertheless, it seems to be clear enough that the large hyperpolarizing action of FC is effective exclusively at the plas-

**Table 4.** Effect of 10<sup>-5</sup> M fusicoccin (FC) on electrical potential differences in leaf slices of *Kalanchoë daigremontiana* 60–90 min after preparation<sup>a</sup>

Before addition of FC:	$-182 < \psi_{co} < -165$	$-160 < \psi_{vo} < -140$
After addition of FC:	$-280 < \psi_{co} < -266$	$-270 < \psi_{vo} < -250$
Difference caused by FC:	$-101 < \Delta\psi_{FC} < -98$	$\Delta\psi_{FC} \approx -110$

<sup>a</sup> Ranges of 18 individual measurements are given in mV

malemma. Taking the data of Table 4 at face value, FC would depolarize the positive potential at the tonoplast by about 10 mV. Although this is almost 50% of the tonoplast potential, in absolute terms it is only a small and not very significant change.

#### K<sup>+</sup>-Transport and Putative H<sup>+</sup> Pumps at the Plasmalemma and Tonoplast of CAM Leaf Cells

It has been postulated that *Kalanchoë* leaf cells have proton pumps at the plasmalemma and tonoplast extruding protons out of the cytoplasm to the external medium and into the vacuole, respectively (Lüttge & Ball, 1979).

Transcellular electrical profiles and earlier uncoupler experiments (Lüttge & Ball, 1979) clearly show that there is a metabolism-dependent electrogenic transport mechanism at the plasmalemma. Evidence that this is an H<sup>+</sup> extrusion pump largely comes from the fusicoccin experiments. In the same way as it has been demonstrated for other higher plant cells, K<sup>+</sup> uptake in exchange for H<sup>+</sup> could be associated with this H<sup>+</sup> extrusion pump (Marré, 1977, 1979; Pitman, Schaefer & Wildes, 1975 a, b).

Evidence for H<sup>+</sup> pumping across the tonoplast into the vacuole mainly rests on an evaluation of intracellular H<sup>+</sup> electrochemical gradients (Lüttge & Ball, 1979). An H<sup>+</sup> pump at the tonoplast could be

responsible for accumulation of malic acid in the vacuoles of the CAM leaves. Exchanging  $K^+$  for  $H^+$ , it could also mediate metabolism-dependent  $K^+$  transport from the vacuole into the cytoplasm. It appears that this mechanism is not electrogenic. It also differs in other respects from the ion pump at the plasmalemma, e.g., it is relatively unaffected by fusicoccin. It is now well established that ATPases, which perhaps pump protons, exist at the tonoplast and other vacuolar membranes (e.g., those of vacuo-lysosomal membranes of *Hevea latex*) (Matile, 1978; Doll, Rodier & Willenbrink, 1979; Guy, Reinhold & Michaeli, 1979; Marin, Marin-Lanza & Komor, 1980).

*In vitro* studies with isolated tonoplasts of CAM leaf cells are urgent to test the existence of a membrane-bound ATPase and its biochemical properties; e.g., its dependence on pH and  $K^+$  and  $Mg^{2+}$  ions. Experiments with intact leaves are required to investigate the possibility of diurnal variations of the intracellular compartmentation of  $K^+$  and  $Mg^{2+}$ , and perhaps other inorganic ions during the CAM rhythm. Taken together, such approaches would allow one to evaluate the idea suggested by the present results, that inorganic ions possibly might be involved in CAM regulation.

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